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Technician's guide to microelectronics
Douglas Young, author
1989
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Technician's Guide to **MICROELECTRONICS**

by John Douglas-Young

From discovering the basic advantages and disadvantages of bipolar and unipolar technologies to troubleshooting the most sophisticated microelectronic equipment—both technicians and engineers will find this guide an immense help in mastering the rapidly-maturing field of microelectronics.

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F Rizvi



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web feb 24 2019 question 1 it s important to be able to recognize the major types of chemical reactions comstock getty images the chemical reaction $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ is a a synthesis reaction b decomposition reaction c single displacement reaction d double displacement reaction e combustion reaction question 2

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web an answer key document has been included and is available for download some additional information to be aware of for teachers demonstrations 1 and 5 will not produce a chemical reaction five of the reactions will produce indicators of a chemical reaction for students to observe

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web identifying chemical reaction types key part 1 identify the type of reaction 1 $\text{Na}_3\text{PO}_4 + 3\text{KOH} \rightarrow 3\text{NaOH} + \text{K}_3\text{PO}_4$ double displacement 2 $\text{MgCl}_2 + \text{Li}_2\text{CO}_3 \rightarrow \text{MgCO}_3 + 2\text{LiCl}$ double displacement 3 $\text{C}_6\text{H}_{12} + 9\text{O}_2 \rightarrow 6\text{CO}_2 + 6\text{H}_2\text{O}$ combustion 4 $\text{Pb} + \text{FeSO}_4 \rightarrow \text{PbSO}_4 + \text{Fe}$ single displacement 5 $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$ decomposition 6 $\text{P}_4 + 3\text{O}_2 \rightarrow 2\text{P}_2\text{O}_3$

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web chemical equations can also be used to represent physical processes write a chemical reaction for the boiling of water including the proper phase labels chemical equations can also be used to represent physical processes write a chemical reaction for the freezing of water including the proper phase labels explain why $4\text{Na} + \text{S} \rightarrow 2\text{Na}_2\text{S}$

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web name score date balancing and identifying chemical reactions a identify each type of chemical reactions 1 $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$ 2 $\text{SrCO}_3 \rightarrow \text{SrO} + \text{CO}_2$ 3 $\text{LiBr} + \text{F}_2 \rightarrow \text{LiF} + \text{Br}_2$ 4 $2\text{Na} + \text{F}_2 \rightarrow 2\text{NaF}$ 5 $\text{Br}_2 + \text{H}_2 \rightarrow 2\text{HBr}$ b balance the chemical reactions and identify each type

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web 1 $3\text{NaBr} + \text{H}_3\text{PO}_4 \rightarrow \text{Na}_3\text{PO}_4 + 3\text{HBr}$ type of reaction double displacement 2 $3\text{Ca}(\text{OH})_2 + \text{Al}_2\text{SO}_4 \rightarrow 3\text{CaSO}_4 + 2\text{Al}(\text{OH})_3$ type of reaction double displacement 3 $3\text{Mg} + \text{Fe}_2\text{O}_3 \rightarrow 2\text{Fe} + 3\text{MgO}$ type of reaction single displacement 4 $\text{C}_2\text{H}_4 + 3\text{O}_2 \rightarrow 2\text{CO}_2 + 2\text{H}_2\text{O}$ type of reaction combustion 5 $2\text{PbSO}_4 + 2\text{PbSO}_3 + \text{O}_2$ type of reaction

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