

"... INSPIRATION AND INSIGHTS FOR ANYONE DETERMINED TO MAKE A DIFFERENCE."

—SECRETARY HILLARY RODHAM CLINTON

Foreword by
**ABRAHAM
VERGHESE**

NO WOMAN LEFT BEHIND

**A Journey of Hope to Heal Every
Woman Injured in Childbirth**

KATE GRANT

Founding CEO, Fistula Foundation

Woman Who Left

Kate Grant



Woman Who Left:

No Woman Left Behind Kate Grant, 2025-06-24 NOW A USA TODAY BESTSELLER INSPIRATION FOR HOW TO CREATE A LIFE OF PURPOSE NO WOMAN LEFT BEHIND IS THE UNLIKELY STORY OF HOW ONE WOMAN LEAVES MADISON AVENUE AND TACKLES THE GLOBAL MATERNAL HEALTH CRISIS HEAD ON PREFACED BY A POWERFUL FOREWORD BY BEST SELLING AUTHOR ABRAHAM VERGHESE 2025 International Book Award Winner in Autobiography Memoir In this rousing debut Grant demonstrates how the faith and actions of a small group of people can save the world BookLife Reviews Editor's Pick A call to action for compassionate advocates and a stunning tale of how deeds triumph over platitudes Dr Denis Mukwege winner of the Nobel Peace Prize The day a woman gives birth is also the day she is most likely to die or suffer severe injury a sobering reality that comes into sharp focus when Kate Grant visits the Addis Ababa Fistula Hospital in Ethiopia's capital There she sees row after row of beds occupied by young women afflicted with obstetric fistula a childbirth injury that leaves them incontinent and too often shunned by their communities modern day lepers She soon learns that surgery is the only way to end their suffering In *No Woman Left Behind* Grant recounts her decision to abandon a promising advertising career and the ups and downs of building Silicon Valley based Fistula Foundation from a modest start up into the global leader in fistula treatment Through vivid firsthand accounts of surgeons toiling in remote corners of Africa and Asia we see inside the fight to restore hope to some of the world's most vulnerable women A compassionate army of donors spanning nearly 70 countries makes such life changing care possible Grant demonstrates the profound power of individual action to change lives at scale since Fistula Foundation takes no government money *No Woman Left Behind* is a compelling personal journey and a how to guide for anyone looking to make a lasting difference in the lives of others 100% of the author's net proceeds from *No Woman Left Behind* will go to Fistula Foundation's Love a Sister program to fund free surgeries for women with childbirth injuries *There's Enough Woman Left to Be Your Lady* Junious Epps, Jr., 2003-03

All Women Are Healers Diane Stein, 1990-04-01 By the study experimentation and practice of natural healing women are changing and charting the future of health care Despite heavy resistance or lack of recognition from patriarchal medicine they are nevertheless making positive changes that will continue and increase Women's emphasis on one to one work practiced in mutual agreement and participation is very different from mechanized and big money medicine and has results and successes far beyond expectations The emphasis on self healing returns health care to the consumer to women's lives and bodies for the first time in centuries The medical system cannot control a movement held in the hands of women though it may try Women are taking control again of healing our daughter right for the first time since the matriarchies and the Inquisition from the Introduction *Folk-dances from Old Homelands* Elizabeth Burchenal, 1922 *The Young Woman's Journal*, 1901 *Women who Leave Violent Relationships* Maryann Syers-McNairy, 1990 **The Woman's Temperance Movement ... a Concise History of the Woman's War on Alcohol, Etc** William C. STEEL, 1874 **The Woman Citizen**

,1918 **The Girl's Own Annual** ,1899 **The American Journal of Obstetrics and Diseases of Women and Children** ,1890 *Novels: Kenelm Chillingly. 1897* Edward Bulwer Lytton Baron Lytton,1897 *The Texas Criminal Reports* Texas. Court of Criminal Appeals,1898 **Popular Educator** ,1919 **Roscoe's Digest of the Law of Evidence in Criminal Cases** Henry Roscoe,Horace Smith,1874 **Words of Weight on the Woman Question** A. H.,1871 **The International Quarterly** ,1902 The Saturday Evening Post ,1917 *Annual Report* Peter Bent Brigham Hospital,1920
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alaskan coast with his uncle their boat sinks the only survivors are chris and a boy named frank who hates chris immediately chris and frank have no radio no flares no food

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web jan 26 2016 the solubility product or ksp is a way of showing how effective a chemical is at dissolving into an aqueous solution the higher the ksp the more the chemical will dissolve into that solvent for this lab calcium hydroxide is created then ionized Ca(OH)_2 when calcium is put in water reacts as follows $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$

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web k sp of calcium hydroxide 1 author j m mccormick last update october 13 2013 introduction an equilibrium constant K is related to ΔG for a process through eqn 1 and as long as ΔH and ΔS are independent of temperature eqn 1 can be combined with eqn 2 to determine ΔH and ΔS from the temperature dependence of K

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web sep 18 2020 in this experiment we will titrate a saturated calcium hydroxide solution with a standardized hcl solution we will determine the hydroxide ion concentration for the saturated calcium hydroxide

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web your primary objective in this experiment is to test a saturated solution of calcium hydroxide and use your observations and measurements to calculate the ksp of the compound you will do this by titrating the prepared Ca(OH)_2 solution with a standard hydrochloric acid solution

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web in today's experiment you will determine the solubility product ksp of calcium hydroxide Ca(OH)_2 by measuring the concentration of Ca(OH)_2 in a saturated solution calcium hydroxide is a sparingly soluble salt that dissolves according to the following reaction the solubility product expression for this reaction is

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web expert verified chem 182 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly

soluble in water a saturated aqueous solution of Ca(OH)_2 is represented in equation form as shown below $\text{Ca(OH)}_2(\text{s}) \rightleftharpoons \text{Ca}^{2+}(\text{aq}) + 2\text{OH}^{-}(\text{aq})$
 the solubility product expression describes in mathematical terms the equilibrium

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web K_{sp} for calcium hydroxide calcium hydroxide is a soft white caustic powder used in making mortar cements calcium salts paints and petrochemicals it is also used in saltwater aquaria to make up kalkwasser limewater solutions for reef tanks and is used as a pH regulating agent notice that calcium hydroxide is divalent and thus has twice

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web use the concentration of calcium hydroxide to calculate the concentration of calcium ions and hydroxide ions in the solution $\text{Ca(OH)}_2 \rightarrow \text{Ca}^{2+} + 2\text{OH}^{-}$ $5 \times 10^{-2} \text{ M}$ calcium ions $5 \times 10^{-2} \text{ M}$ 0 M hydroxide ions calculate calcium hydroxide K_{sp} using the concentrations of calcium ions and hydroxide ions

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web madison bowe elana keller april 15 2014 introduction the objective of this lab was to measure the K_{sp} of a slightly soluble basic salt and how solubility changes with temperature K_{sp} is another form of equilibrium expression that deals with solubility the reaction that will take place is $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^{-}$

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web question experiment 8 determining K_{sp} for calcium hydroxide post lab report sheet data table 2 points trial 1 trial 2 HCl molarity 0.050 M volume of filtered Ca(OH)_2 solution ml 5.00 5.00 3.71 13.75 equivalence point volume ml OH⁻ in saturated Ca(OH)_2 M Ca²⁺ in saturated Ca(OH)_2 M K_{sp} of Ca(OH)_2 average K_{sp} of Ca(OH)_2

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web the chemical equation used was $\text{Ca(OH)}_2(\text{s}) \rightleftharpoons \text{Ca}^{2+}(\text{aq}) + 2\text{OH}^{-}(\text{aq})$ the solubility product was determined to be K_{sp} $\text{Ca}^{2+}\text{OH}^{-}$ from the titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

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web jul 12 2023 because the concentration of a pure solid such as Ca_3PO_4 is a constant it does not appear explicitly in the equilibrium constant expression the equilibrium constant expression for the dissolution of calcium phosphate is therefore $K = [\text{Ca}^{2+}]^3 [\text{PO}_4^{3-}]$ Ca_3PO_4 $K = [\text{Ca}^{2+}]^3 [\text{PO}_4^{3-}]$

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web lab report 8 determination of calcium hydroxide the purpose of this was to titrate a saturated Ca(OH)_2 solution with a standard HCl solution and find the K_{sp} or the OH⁻ concentration was found and then the concentration was found and used for this experiment a titration of calcium hydroxide with hydrochloric

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web question determining the ksp of calciumhydroxide Ca(OH)_2 s Ca^{2+} aq 2OH^- aq for this experiment the titration of calciumhydroxide with hydrochloric acid was investigated determining the ksp of calcium hydroxide Ca(OH)_2 s Ca^{2+} aq 2OH^- aq for this experiment the titration of calcium

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web question chem 182 experiment 8 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water solution of Ca(OH)_2 is represented in equation form lution of Ca(OH)_2 ionic solid that is sparingly soluble in water a saturated aqueous as shown below

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web therefore it was determined through error calculations and ease of use that the most reliable valid method to use to get the closest ksp value to the theoretical ksp for Ca(OH)_2 was the determination through ph method 2 introduction calcium hydroxide is a strong base soluble in water

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web plasma membrane homework symbiosis practice worksheet benzoin condensation grade a determining the ksp of calcium hydroxide after reading the lab complete items a b c and d title purpose chemicals and equipment and summary of procedure as described on page 10 of exp 1 on an 8 1/2 x 11 sheet of paper

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web question lab 9 determination of the solubility product ksp of calcium hydroxide introduction ionic compounds that are classified as insoluble based on solubility rules are actually slightly soluble each of these insoluble compounds actually dissolves to

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web we will filter the solid calcium hydroxide from the solution and then titrate the dissolved calcium hydroxide in solution with a known concentration of hydrochloric acid using titrimetric calculations we will determine the Ca^{2+} and OH^- concentrations in solution

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web calculate the ksp for calcium hydroxide Ca(OH)_2 $K_{sp} = 5 \times 10^{-6}$ find the accepted value of the ksp for calcium hydroxide and compare it with your value discuss the discrepancy and suggest possible sources of experimental error accepted value for ksp for sodium hydroxide is 5×10^{-6}

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web in this lab you will experiment with a half life model in which m m candies represent radioactive atoms the imprinted m on each candy represents whether the atom has

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web the exponential law can also be interpreted as the decay probability for a single radioactive particle to decay in the interval dt about t this probability p t properly

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web decay 1 parent isotope parent isotope decay radioactive daughter t radioactive daughter t dt decay 1 decay 2 dt init radioactive daughter

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web jan 10 2021 although the radioactive decay of a nucleus is too small to see with the naked eye we can indirectly view radioactive decay in an environment called a cloud

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web the decay of radioactivity in a radioactive element can be modelled using cubes dice or coins in decay a radioactive parent nucleus randomly emits an alpha or beta particle

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web this project involved the simulation of radioactive decay the probability p that a certain atom will decay in a small

amount of time t is $p \lambda t = 1 - \lambda \ln 2 t^{1/2}$ where λ

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web the most common types of radioactivity are α decay β decay γ emission positron emission and electron capture nuclear reactions also often involve γ rays and some

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web during alpha decay an element emits a helium nucleus with 2 neutrons and 2 protons thus the atomic mass of the new element is decreased by four and the atomic number

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web today we know that alpha rays are helium nuclei beta rays are high energy electrons and gamma rays are high energy photons particles of light certain isotopes of

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web nov 21 2021 the equation below describes the radioactive decay of a uranium 238 atom where a thorium 234 atom and an alpha particle are formed ${}_{92}^{238}\text{U} \rightarrow {}_{90}^{234}\text{Th} + {}_2^4\text{He}$

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web after harvesting a radioactive sample from a fallen meteor you will start with the fundamentals learning about the atom stability and the origins of radioactive decay

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