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Knaive*

USA TODAY-BESTSELLING AUTHOR

Patricia Potter

# The Black Knave

**Albert A Gayle**



**The Black Knave:**

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web use crayons or colored pencils to color in the spectrum in the box on your report sheet to show the correlation between

color and wavelength part ii flame tests you will conduct flame tests to observe the flame emission colors for the following solutions  $\text{LiNO}_3$   $\text{CuNO}_3$   $2\text{SrNO}_3$   $2\text{BaNO}_3$   $2\text{KNO}_3$  and  $\text{NaNO}_3$  since nitrates do not emit color you

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metal cations in this lab how do the metal cations become excited

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web feb 10 2021 section 1 purpose and summary observe the color of light emitted when a sample of a salt of a metal cation is heated use the characteristic color of flames to identify ions in various laboratory samples observe the emission spectra of various elements in gas discharge tubes

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