

Periodic Table

These are the 114 currently known and officially named elements that comprise the periodic table (IUPAC, 2016). Each element is represented by an image produced for the Royal Society of Chemistry project.

The periodic table arranges the elements, such that those sharing physical and chemical properties, or those of similar behavior, and the three main groups of elements (hydrogen, alkali metals, and the noble gases) are grouped with closely related physical properties. Related elements, in order of mass, and are related to their periodicity. Properties of elements change in a systematic way through a period.

Atomic number
The atomic number is equal to the number of protons in the nucleus.

Relative atomic mass
The ratio of the average mass of the element to the mass of a proton is one-twelfth of the mass of a proton of about 1.66 x 10⁻²⁷ kg. A number in brackets indicates that all isotopes of the element are unstable, or radioactive.

Transition metals

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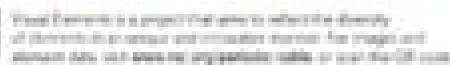
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