



Teaching In The Congregation

RJ Shavelson



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web an solution initially contains a catalytic amount of an enzyme with $k_m = 1.5 \times 10^{-5} \text{ M}$ of substrate and no product after 45 seconds the solution contains $25 \mu\text{M}$ of product find v_{max} and the concentration of product after 20 minutes hint $s = k_m$

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web a reaction and its experimental rate law are represented below $2 \text{ ICl}_g + \text{H}_2 \text{ g} \rightarrow 2 \text{ HCl}_g$ rate $k [\text{ICl}]^2$ a chemist proposes a mechanism for the reaction that is consistent with the rate law the mechanism has two elementary steps and the first step is slow compared to the second

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web in order to have successful collisions the colliding particles must have both the proper amount of energy and the proper fill in blank draw an energy diagram which shows the shape of the curve you might expect for the reaction in this question the overall reaction is exothermic

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web reaction rates google classroom $\text{mg s} + 2 \text{hcl} \rightarrow \text{mgcl}_2 + \text{h}_2\text{g}$ the reaction between mg s and hcl is represented by the equation above in a kinetics experiment a 0.080 g sample of mg s ribbon is added to an erlenmeyer flask containing 50 ml of 3.0 M hcl at 25 °C

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