

Periodic Table

These are the 118 currently known and officially named elements that comprise the periodic table (IUPAC, 2016). Each element is represented by an image produced for the Visual Elements project.

The periodic table arranges the elements, with their atomic physical and chemical properties, in order of atomic number and by their electronic configuration. It groups elements into the s-block, d-block, p-block and f-block, which are closely related physical properties. However, elements in order of their cell are listed below in groups. Properties of elements change in a systematic way through a period.

Atomic number
The atomic number is equal to the number of protons in the nucleus.

Relative atomic mass
The ratio of the average mass of the atoms to the mass of an atom of carbon-12, which is by ground state. A number of atoms, and even that all isotopes of the element are specified, is indicated.



Group 1
Hydrogen
Group 2

Transition metals

Group 3 Group 4 Group 5 Group 6 Group 7 Group 8 Group 9 Group 10 Group 11 Group 12

Group 13 Group 14 Group 15 Group 16 Halogens Group 17 Group 18

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Project
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Visual Elements is a project that aims to reflect the diversity of elements in an original and innovative manner. For images and element data visit www.rsc.org/periodic-table or scan the QR code.

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