

CALIFORNIA SOCIAL STUDIES Standards

First Grade Color-Coded Focus Wall Cards

Describe similarities and differences of various geographical areas based on their location and within the natural, human, economic, climate, and political, drawing from diagrams, and histories, and features.

HSS-1.4.3



Students describe the human characteristics of familiar places and the varied backgrounds of American citizens and residents in those places.

HSS-1.5



Describe how various parts of the same geographic area, such as mountains, rivers, and oceans, affect the lives of people in their various ways and challenges of a region.

HSS-1.1



Describe the rights and responsibilities of citizenship.

HSS-1.1



Describe the process of a direct democracy or the basic and representative forms of government and how people make their voices heard in their decisions, school, and community.

HSS-1.1.1



Describe elements of fair play and good conduct for the rights and liberties expected for rules by which we live, including the meaning of the "Golden Rule".

HSS-1.1.2



Construct a simple map, using cardinal directions and map symbols.

HSS-1.2.3



Describe human, economic, and physical characteristics of the way people live, including their food, clothing, shelter, education, and recreation.

HSS-1.2.4



Describe and understand the symbols, signs of the United States that identify and a sense of community across time.

HSS-1.3



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the loss of electrons resulting in a more positively charged species reduction the gain of electrons resulting in a

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web we can balance it for charge by adding two electrons to the right side of the equation so that the net charge on each side is 0 oxidation $\text{Ni} \rightarrow \text{Ni}^{2+} + 2\text{e}^-$ now that the oxidation half reaction is balanced it tells us that two electrons are produced for every atom of nickel oxidized but where do those electrons go

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web balancing redox reactions in acidic solution problems 1 10 problem 1 $\text{Cr}_2\text{O}_7^{2-} + \text{Fe}^{2+} + \text{Cr}^{3+} + \text{Fe}^{3+}$ solution 1 balanced half reactions $6\text{e}^- + 14\text{H}^+ + \text{Cr}_2\text{O}_7^{2-} \rightarrow 2\text{Cr}^{3+} + 7\text{H}_2\text{O}$ $\text{Fe}^{2+} \rightarrow \text{Fe}^{3+} + \text{e}^-$ 2 equalize the electrons $6\text{e}^- + 14\text{H}^+ + \text{Cr}_2\text{O}_7^{2-} \rightarrow 2\text{Cr}^{3+} + 7\text{H}_2\text{O}$ $6\text{Fe}^{2+} \rightarrow 6\text{Fe}^{3+} + 6\text{e}^-$ multiplied by a factor of 6

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web in the redox equation below that occurs in our stomach which of the following pairs identifies the reducing and oxidizing agents respectively $\text{C}_6\text{H}_8\text{O}_6$ $2\text{H}_2\text{NO}_2$ $\text{C}_6\text{H}_6\text{O}_6$ $2\text{H}_2\text{O}_2$ 2NO choose 1 answer $\text{C}_6\text{H}_8\text{O}_6$ and 2NO a $\text{C}_6\text{H}_8\text{O}_6$ and 2NO_2 b $\text{C}_6\text{H}_8\text{O}_6$ and $\text{C}_6\text{H}_6\text{O}_6$ c $\text{C}_6\text{H}_8\text{O}_6$ and $2\text{H}_2\text{O}_2$ d $\text{C}_6\text{H}_8\text{O}_6$ and $2\text{H}_2\text{NO}_2$

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