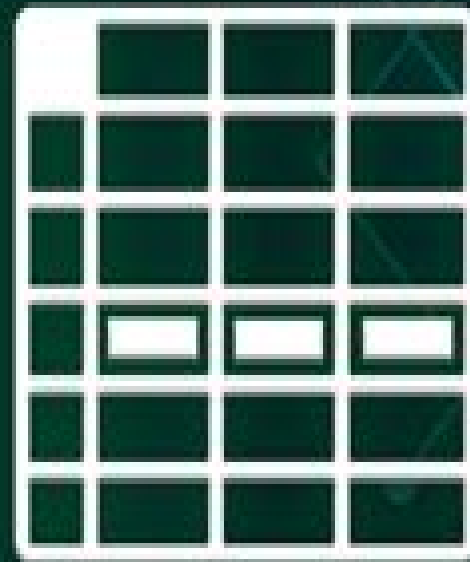


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 $2\ 3\ \text{Ca(OH)}_2 + 1\ \text{Al}_2\text{SO}_4 \rightarrow 3\ \text{CaSO}_4 + 2\ \text{Al(OH)}_3$ type of reaction double displacement
 $3\ 3\ \text{Mg} + 1\ \text{Fe}_2\text{O}_3 \rightarrow 2\ \text{Fe} + 3\ \text{MgO}$ type of reaction single displacement
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