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R Barnett



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web calorimetry worksheet 1 if 0.315 moles of hexane C_6H_{14} is combusted in a bomb calorimeter containing 5.65 liters of water calculate the molar heat of combustion of hexane if the water temperature rises 55.4 °C the specific heat capacity of water is 4.184 J/g°C $m_s t_h = 5.650 \text{ grams } \text{H}_2\text{O} \cdot 4.184 \text{ J/g}^\circ\text{C} \cdot 55.4^\circ\text{C} = 1310 \text{ kJ}$

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web 6 a 218 g sample of steam at 121oc is cooled to ice at 14oc find the change in heat content of the system 7 in going from ice at 34oc to steam at 138oc a sample of water absorbs 1.41×10^5 j find the mass of the sample answers 1.268×10^4 j 2.623×10^4 j 3.171×10^4 j 4.111×10^5 j 5.211×10^5 j 6

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