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# Tt Rivers

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## **Tt Rivers:**

*Birds of the Major Mainland Rivers of Southeast Alaska* James Andrew Johnson, Brad Alan Andres, John A. Bissonette, 2008  
This publication describes the bird communities of major mainland rivers of southeast Alaska and is based on a review of all known relevant studies as well as recent fieldwork We synthesized information on the composition structure and habitat relationships of bird communities at 11 major mainland rivers Information on current management concerns and research needs are also included  
California. Court of Appeal (2nd Appellate District). Records and Briefs California (State)., Court of Appeal Case s Consolidated Case s Number of Exhibits 0  
Proceedings of the American Society of Civil Engineers American Society of Civil Engineers, 1929 Vols for Jan 1896 Sept 1930 contain a separately page section of Papers and discussions which are published later in revised form in the society s Transactions Beginning Oct 1930 the Proceedings are limited to technical papers and discussions while Civil engineering contains items relating to society activities etc

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**Catalogue of the Royal Institute of British Architects** Royal Institute of British Architects, 1889 Archaeologia, Or, Miscellaneous Tracts Relating to Antiquity ,1906 **New York State Education Department Bulletin** ,1916

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**Parliamentary Papers** Great Britain. Parliament. House of Commons, 1913

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## Tt Rivers Introduction

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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) \rightarrow \text{C}_2\text{H}_2(\text{g}) + \frac{5}{2} \text{O}_2(\text{g})$   $\Delta H = 94.5 \text{ kJ}$   
 $\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g})$   $\Delta H = 71.2 \text{ kJ}$   
 $\text{C}_2\text{H}_6(\text{g}) \rightarrow \frac{7}{2} \text{O}_2(\text{g}) + 2 \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{g})$   $\Delta H = 283 \text{ kJ}$

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 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 2 \text{H}_2\text{O}(\text{l})$   $\Delta H = 1411 \text{ kJ/mole}$   
 $\text{C}_2\text{H}_6(\text{g}) + 7/2 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{l})$   $\Delta H = 1560 \text{ kJ/mole}$   
 $\text{H}_2(\text{g}) + 1/2 \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l})$   $\Delta H = 285 \text{ kJ/mole}$

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web thus for reaction 9 2 in which 1 mole of h<sub>3</sub>o and 1 mole of oh combine to form 2 moles of h<sub>2</sub>o h 55 8 kj as shown note that the enthalpy change h<sub>products</sub> h<sub>reactants</sub> is positive if heat is absorbed that is if h<sub>products</sub>

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