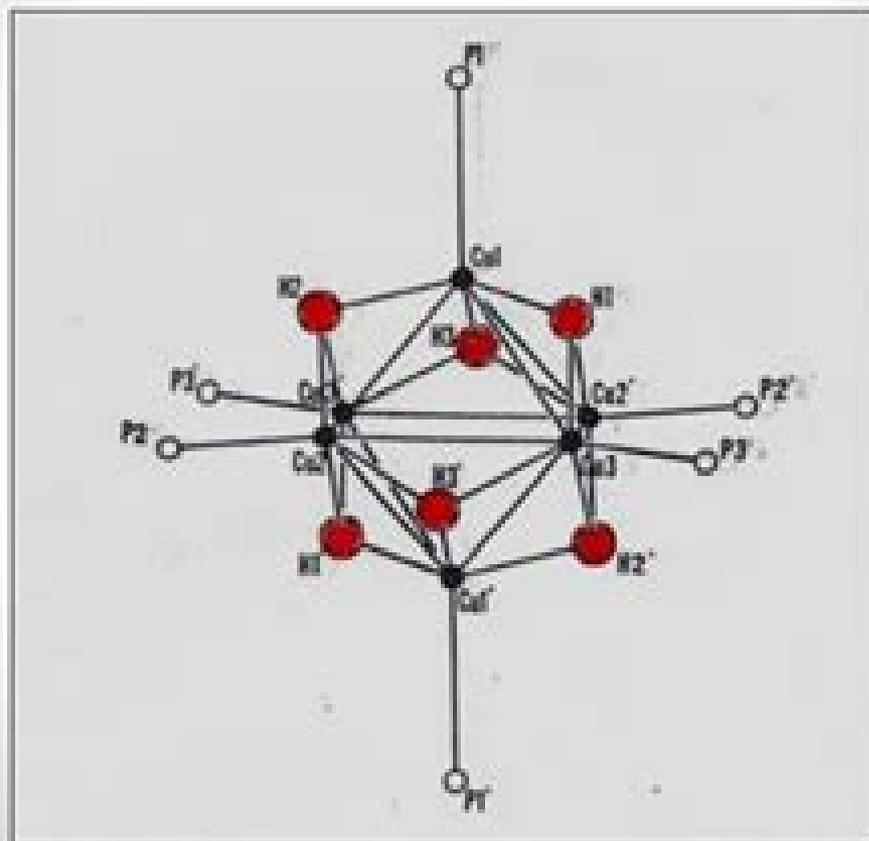


Transition-Metal Hydrides



Stevens, R. C. et al. *J. Am. Chem. Soc.* **1989**, *111*, 3472-3473.

Transition Metal Hydrides

Robert Bau



Transition Metal Hydrides:

Transition Metal Hydrides Alain Dedieu, 1992 **Transition Metal Hydrides** Earl L. Muetterties, 1971 **THE TRANSITION METAL HYDRIDE COMPLEXES.** BERNARD. SIEGEL, AEROSPACE CORP EL SEGUNDO CALIF., 1961 The status of present day knowledge concerning transition metal hydrides that are stabilized by ligand molecules is summarized Particular emphasis is given to the structures and chemical properties of such complex molecules and to conjectures about the possible structures of complexes as yet unknown The literature survey on which this report is based was intended to provide background information for the experimental program on transition metal hydrides Author *Transition Metal Hydrides*, 1978 *Transition Metal Hydrides* Robert Bau, 1978 **Studies on Some Transition Metal Hydrides** Melodye Wood Marks, 1981 *Catalytic Transition Metal Hydrides* Donald Warren Slocum, William R. Moser, 1983

Thermodynamic Hydricity of Transition Metal Hydrides, 2016 Transition metal hydrides play a critical role in stoichiometric and catalytic transformations Knowledge of free energies for cleaving metal hydride bonds enables the prediction of chemical reactivity such as for the bond forming and bondbreaking events that occur in a catalytic reaction Thermodynamic hydricity is the free energy required to cleave an M-H bond to generate a hydride ion H⁻ Three primary methods have been developed for hydricity determination the hydride transfer method establishes hydride transfer equilibrium with a hydride donor-acceptor pair of known hydricity the H₂ heterolysis method involves measuring the equilibrium of heterolytic cleavage of H₂ in the presence of a base and the potential pK_a method considers stepwise transfer of a proton and two electrons to give a net hydride transfer Using these methods over 100 thermodynamic hydricity values for transition metal hydrides have been determined in acetonitrile or water In acetonitrile the hydricity of metal hydrides spans a range of more than 50 kcal/mol Finally methods for using hydricity values to predict chemical reactivity are also discussed including organic transformations the reduction of CO₂ and the production and oxidation of hydrogen Some Reactions of Transition Metal Hydrides B. L. Shaw, Chemistry, 1969 **Organotransition Metal Chemistry** Anthony F. Hill, 2002 This book aims to introduce undergraduates to the utility of organotransition metal chemistry a discipline of importance to scientists in a variety of industry sectors *Transition Metal Hydride Complexes from Protonation Reactions: Formation and Reactivity* David Wayne Mosley, 1997 **Transition Metal Hydrides of the Iron Nitrosyl Group** Alain Forgues, 1981 *Recent Advances in Hydride Chemistry* Maurizio Peruzzini, Rinaldo Poli, 2002-01-30 This book has been conceived to collect the most important recent advances in all areas of hydride chemistry research including chemical reactivity instrumental investigation theory and applications in the areas of catalysis biochemistry and materials science Many of the chapters have been written by the plenary lecturers of the EURO Hydrides 2000 conference but other leading scientists in this field have also been invited to contribute The first part of the book focuses on the chemistry and catalysis of transition metal hydrides Another block of chapters illustrates the most recent advances in the application of instrumental

techniques to the study of the properties and reactivity of hydride compounds The final part of the book illustrates the relevance of metal hydrogen bonds in biochemistry and materials science All of the chapters of this book have been evaluated by independent reviewers

Transition Metal Hydrides as Reagents in Organic and Organometallic Synthesis Kay Alexandra Youngdahl,1987

Metal Hydrides Gust Bambakidis,2013-11-11 In the last five years the study of metal hydrides has expanded enormously due to the potential technological importance of this class of materials in hydrogen based energy conversion schemes The scope of this activity has been worldwide among the industrially advanced nations There has been a consensus among researchers in both fundamental and applied areas that a more basic understanding of the properties of metal hydrogen systems is required in order to provide a rational basis for the selection of materials for specific applications The current worldwide need for and interest in research in metal hydrides indicated the timeliness of an Advanced Study Institute to provide an in depth view of the field for those active in its various aspects The inclusion of speakers from non NATO countries provided the opportunity for cross fertilization of ideas for future research While the emphasis of the Institute was on basic properties there was a conscious effort to stimulate interest in the application of metal hydrides to solar hydrogen energy conversion schemes in land areas where solar energy has promise as a primary energy source In addition to the lectures several seminars were given which treated topics of special interest in greater detail

Transition metal hydrides : proceedings Robert Bau,1978

COMPLEX TRANSITION METAL HYDRIDES. S. G. Gibbins,AEROSPACE CORP EL SEGUNDO CALIF.,1963 A review of the properties of complex transition metal hydrides and a discussion of the probable mechanism requisite for stability are given The absence of stoichiometric binary transition metal hydrides coupled with features common to the complex ones indicates that the primary stabilization mechanism of the transition metalhydrogen bond is one involving multiple bonding between the metal and the stabilizing ligand The multiple bond comprises of a sigma bond the electron pair being provided by the ligand and a pi bond the electron pair being provided by the metal Thus negative charge that would have accumulated on the metal if only a sigma bond existed is diminished From a molecular orbital point of view the pi bond permits formerly metal non bonding d orbitals to become bonding Thus the total energy of the system is reduced and the stability is increased Author

Transition Metal Hydrides Maria Margaret Ludvig,1987

Basic Organometallic Chemistry Ionel Haiduc,Jerry J. Zuckerman,2011-06-01 No detailed description available for Basic Organometallic Chemistry

Transition Metal Hydrides Robert Bau,1978

Transition Metal Hydrides Book Review: Unveiling the Power of Words

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