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**Emmerich Wilhelm, Trevor M Letcher**



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Standard Reference Data National Measurement Laboratory (U.S.). Office of Standard Reference Data, Herman M. Weisman, Gertrude B. Sherwood, 1970      **Bulletin of Thermodynamics and Thermochemistry**, 1978      **Bulletin of Chemical Thermodynamics**, 1989      *Volume Properties* Emmerich Wilhelm, Trevor M Letcher, 2014-11-25 Volumetric properties play an important role in research at the interface of physical chemistry and chemical engineering but keeping up with the latest developments in the field demands a broad view of the literature Presenting a collection of concise focused chapters this book offers a comprehensive guide to the latest developments in the field and a starting point for more detailed research The chapters are written by acknowledged experts covering theory experimental methods techniques and results on all types of liquids and vapours The editors work at the forefront of thermodynamics in mixtures and solutions and have brought together contributions from all areas related to volume properties offering a synergy of ideas across the field Graduates researchers and anyone working in the field of volumes will find this book to be their key reference

Thermodynamics and the Free Energy of Chemical Substances Gilbert Newton Lewis, Merle Randall, 1923 The scope of thermodynamics Definitions the concept of equilibrium Conventions and mathematical methods Solutions The first law of thermodynamics and the concept of energy The fugacity Application of the second law to solutions The perfect solution The laws of the dilute solution Systems involving variables other than pressure temperature and composition A useful function called the activity and its application to solutions Change of activity with the temperature and the calculation of activity from freezing points The standard change of free energy the equilibrium constant Solutions of electrolytes The activity of strong electrolytes The activity of electrolytes from freezing point data and tables of activity coefficients Activity coefficient in mixed electrolytes the principle of the ionic strength the activity of individual ions The galvanic cell Single potentials standard electrode potentials of the elements The third law of thermodynamics The entropy of monatomic gases and a table of atomic entropies Introduction to systematic free energy calculations the free energy of elementary hydrogen and metallic hydrides Oxygen and its compounds with hydrogen and with some metals Chlorine and its compounds Bromine and its compounds Iodine and its compounds Nitrogen compounds Carbon and some of its compounds Compounds of carbon and nitrogen Table of free energies and examples illustrating its use Conversion table for mol fractions mol ratios and molities Some useful numerical factors Coefficients employed in converting activity equilibrium constant and free energy from one temperature to another Publications by the authors pertaining to thermodynamics      Handbook of Viscosity: Volume 1 Carl L. Yaws, 1995 This series provides engineers with liquid and gas viscosities for the major organic compounds as a function of temperature The graphs are arranged by chemical formula to provide ease of use many of them cover the full range from melting point to boiling point to critical point Common units are used but each graph displays a conversion factor to provide English units

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