

pH PERFECT® CONNOISSEUR® MASTER RECIPE GLOBAL

GROW CYCLE


18 HRS OF LIGHT
6 HRS OF DARK

	WEEK 1	WEEK 2	WEEK 3	WEEK 4
 pH PERFECT CONNOISSEUR GROW A	1 mL/L	2 mL/L	3 mL/L	4 mL/L
 pH PERFECT CONNOISSEUR GROW B	1 mL/L	2 mL/L	3 mL/L	4 mL/L
 VOOODOO JUICE®	2 mL/L	2 mL/L		
 TARANTULA®	2 mL/L	2 mL/L		
 PIRANHA®	2 mL/L	2 mL/L		
 RHINO SKIN®	2 mL/L	2 mL/L	2 mL/L	2 mL/L
 B-52®	2 mL/L	2 mL/L	2 mL/L	2 mL/L
 SENSIZYM®	2 mL/L	2 mL/L	2 mL/L	2 mL/L
 BUD CANDY®	2 mL/L	2 mL/L	2 mL/L	2 mL/L

LEGEND

-  BASE NUTRIENTS
-  ROOT EXPANDERS
-  BUD POTENCY & STALK STRENGTHENERS
-  BIGGER BUDS
-  GROW MEDIUM CONDITIONER
-  BUD TASTE & TERPENE ENHANCEMENT

BLOOM CYCLE


12 HRS OF LIGHT
12 HRS OF DARK

	WEEK 1	WEEK 2	WEEK 3	WEEK 4	WEEK 5	WEEK 6	WEEK 7	WEEK 8
 pH PERFECT CONNOISSEUR BLOOM A	4 mL/L							
 pH PERFECT CONNOISSEUR BLOOM B	4 mL/L							
 VOOODOO JUICE®	2 mL/L	2 mL/L						
 TARANTULA®	2 mL/L	2 mL/L						
 PIRANHA®	2 mL/L	2 mL/L						
 BUD FACTOR X®	2 mL/L							
 RHINO SKIN®	2 mL/L							
 B-52®			2 mL/L					
 NIRVANA®			2 mL/L					
BUD IGNITOR®	2 mL/L	2 mL/L						
BIG BUD®		2 mL/L	2 mL/L	2 mL/L	2 mL/L			
OVERDRIVE®						2 mL/L	2 mL/L	
SENSIZYM®	2 mL/L							
BUD CANDY®	2 mL/L							
FLAWLESS FINISH®								2 mL/L

NUTRIENT FREE FLUSH PERIOD



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Jicheng Xie



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web $m \times c \times \Delta t$ equation 1 where m is the total mass of the solution solute plus solvent c is the specific heat of the solution

and Δt is the observed temperature change the specific heat of the solution is generally assumed to be the same as that of
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web the heat of reaction ΔH_{soln} is written after the products in units of kJ mol rxn for example sodium hydroxide dissolves exothermically releasing 44.2 kilojoules per mole dissolved this process is represented as $\text{NaOH (s)} \rightarrow \text{NaOH (aq)}$ $\Delta H_{\text{soln}} = -44.2 \text{ kJ/mol rxn}$ write equations to similarly represent the dissolving process for

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web the goal of my experiment and research is to evaluate the various number of compounds to be used as hand warmers and locate the ones that can raise the temperature to 20 degrees celsius in 100 g water and costs as little as possible

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web in this lab you are challenged to use chemistry to determine which ionic compound makes the best hand warmer in this experiment you will collect data that will allow you to calculate the heat of solution ΔH_{soln} or the amount of energy released or absorbed in kJ/mol of the substance being dissolved

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