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web we utilize the equation that relate amount remaining initial mass and number of half lives n n t 1 2 n x n o 1 2 4 x 50 3 125 g 5 what is the half life of an isotope that is 80 remained after 16 days solution remaining 80 100 therefore n t 80 n o 100 now using the half life equation in example 1 we have

web what is the half life of this radioisotope after 2 days 5 grams of the radioisotope 1 day b 2 days c 5 days d 4 days 3 the ratio of uranium 238 to lead 206 in a mineral is used to determine age b density c solubility d composition 4 a radioactive isotope has a half life of 10 years

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using the previous equation n 4 substituting and solving results in the following less than one third of a gram of 20 f remains skill building exercise the half life of 44 ti is 60 0 y

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