

THE SANGUINE SURRENDER



# The Sanguine Surrender

**CL Gary**



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web 1 molarity mv mass molar mass  $\times 0.4500 \text{ L} \times 0.825 \text{ g/L} \times 0.0129 \text{ mol/L} = 0.00581158 \text{ mol}$  2 molality  $0.825 \text{ g/L} \times 0.0129 \text{ mol/L} = 0.0106425 \text{ mol/kg}$  3 mole fraction  $n_2 = 0.00581158 \text{ mol}$   $n_1 = 0.4500 \text{ mol} - 0.00581158 \text{ mol} = 0.44418841 \text{ mol}$

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