

Valency of First 20 Elements

Element Name	Symbol	Atomic Number	Electronic Configuration	Lose/gain/share Electrons	Valency
Hydrogen	H	1	1	Lose or share 1 Electron	1
Helium	He	2	2	Stable as per Duplet rule	0
Lithium	Li	3	2,1	Lose 1 Electrons	1
Beryllium	Be	4	2,2	Lose 2 Electrons	2
Boron	B	5	2,3	Lose 3 Electrons	3
Carbon	C	6	2,4	Share 4 Electrons	4
Nitrogen	N	7	2,5	Gain or share 3 Electrons	3
Oxygen	O	8	2,6	Gain or share 2 Electrons	2
Fluorine	F	9	2,7	Gain or share 1 Electrons	1
Neon	Ne	10	2,8	Stable as per Octet rule	0
Sodium	Na	11	2,8,1	Lose 1 Electrons	1
Magnesium	Mg	12	2,8,2	Lose 2 Electrons	2
Aluminum	Al	13	2,8,3	Lose 3 Electrons	3
Silicon	Si	14	2,8,4	Share 4 Electrons	4
Phosphorous	P	15	2,8,5	Gain or share 3 Electrons	3
Sulfur	S	16	2,8,6	Gain or share 2 Electrons	2
Chlorine	Cl	17	2,8,7	Gain or share 1 Electrons	1
Argon	Ar	18	2,8,8	Stable as per Octet rule	0
Potassium	K	19	2,8,8,1	Lose 1 Electrons	1
Calcium	Ca	20	2,8,8,2	Lose 2 Electrons	2

Valency And Molecular Structure

B Lingard



Valency And Molecular Structure:

Valency And Molecular Structure Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Valency And Molecular Structure**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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