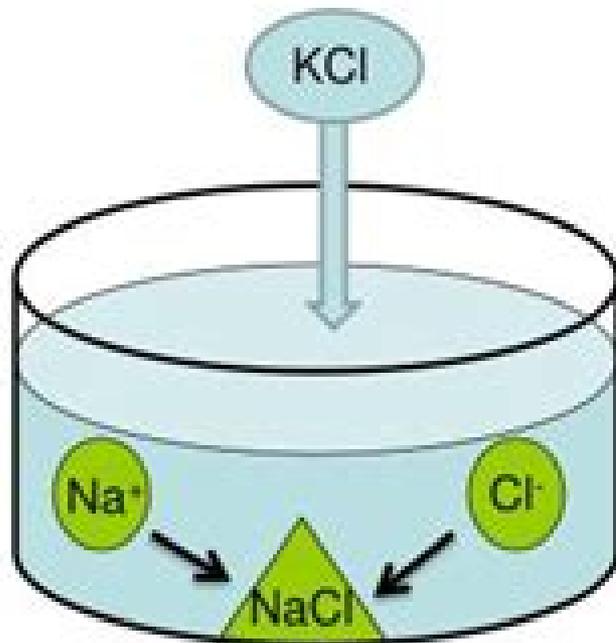


Common Ion Effect

- With solutions that are saturated, addition of either ion, from another compound, will shift the equilibrium



To a saturated NaCl solution, add KCl
Cl⁻ concentration goes up
Equilibrium shifts left
Some NaCl will precipitate
Na⁺ concentration goes down

The Ion Effect

Brendan G. Carr



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