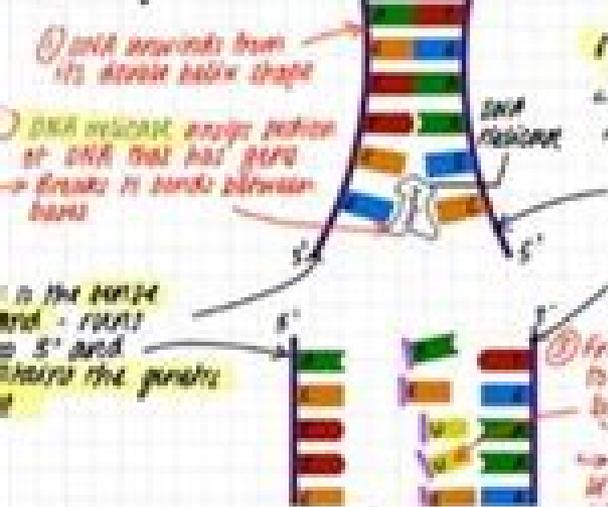


Description:



Hydrogen bonding
 - weak intermolecular forces between H₂O and F₂O (or, O, F) that has a lone pair of e⁻.



LIQUID
 - Provides a reaction medium for chemical reactions (e.g. cytoplasm)
 - Provides efficient transport medium

High surface area to volume ratio
 - Provides a suitable habitat environment for aquatic conditions
 - Provides control temperature for enzyme catalysed reactions
 - If bonds require greater amounts of energy to overcome its bonds due to increased density.

LIQUID GREAT HEAT OF VAPORIZATION

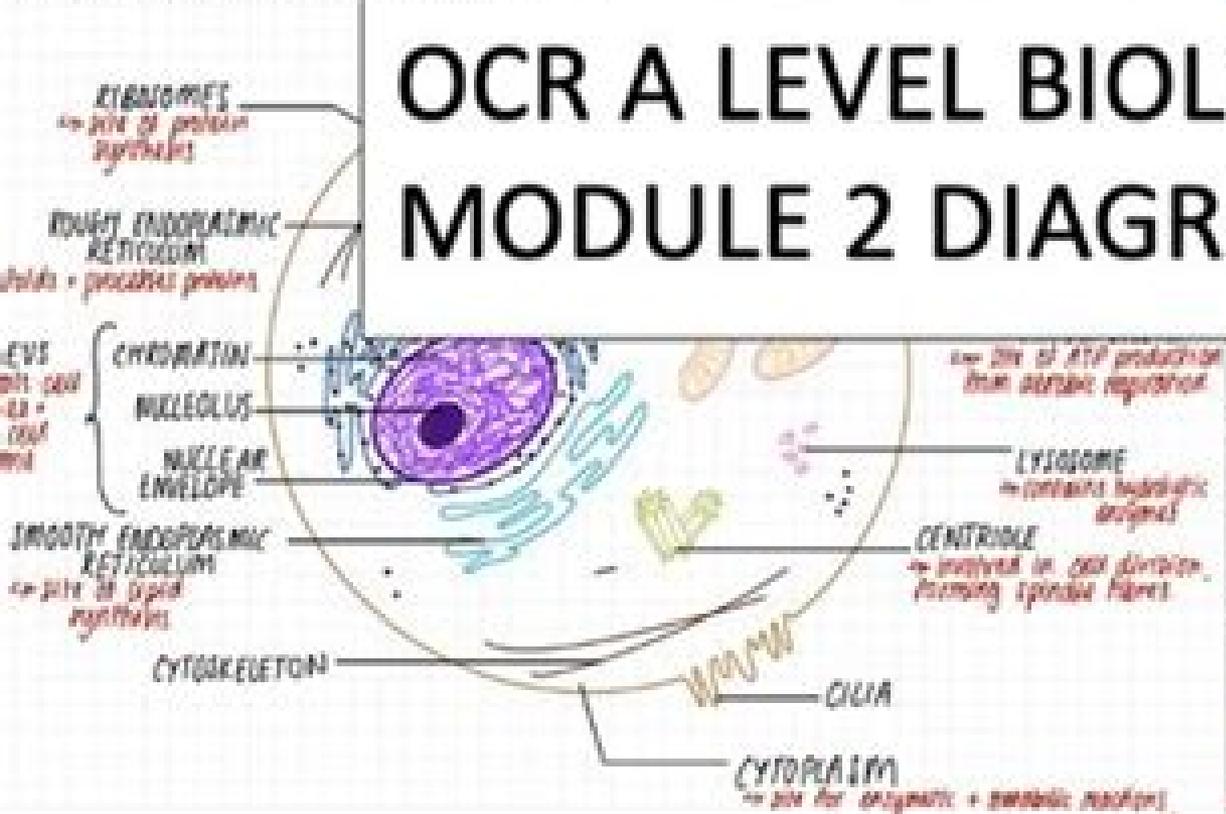
- Provides a cooling effect
 - e.g. sweat evaporation at animal's body surface
 - Large amount of energy is needed to overcome its bonds
 - Large amount of energy released when bonds break

LIQUID
 - Provides slight warmer environment for aquatic organisms in the winter
 - Heat loss is slow cause of liquid water so the heat and organisms water loss
 - In solid ice, H bonds to other water molecules to form lattice structure to not break.

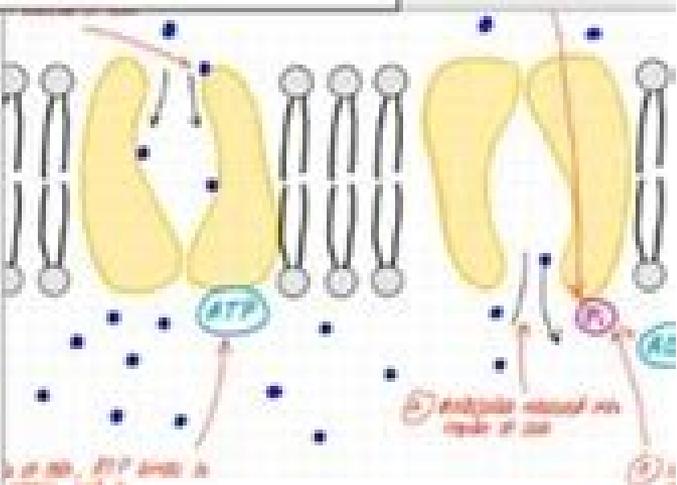
CONDENSATION, ADSORPTION + SURFACE TENSION

- Cohesive - adhesive properties ensure transportation through plant xylem up to leaf
 - Surface tension - enables a fish to pond surface to walk on pond surface to

Animal Cell Ultrastructure:



OCR A LEVEL BIOLOGY MODULE 2 DIAGRAMS



Osmosis - Attraction between water molecules due to H₂O
Diffusion - Attraction between water molecules and other small solute molecules due to H₂O bonds
Water Potential - Potential in water due to concentration of molecules and pressure due to H₂O.

Statistics 2 For Ocr

Weston Flint



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Statistics 1 for OCR Steve Dobbs, Jane Miller, 2000-06-15 This book is part of the Cambridge Advanced Level Mathematics series written specifically for the OCR modular mathematics syllabus Each book or half book corresponds exactly to one module Statistics 1 has chapters on data representation probability expectation and variance and the binomial distribution It also introduces correlation and regression Each chapter starts with a short list of learning objectives and mathematical ideas are explained carefully and clearly with many worked examples There are plenty of exercises throughout along with revision exercises and mock exam papers all written by experienced examiners

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