

Combustion Processes & Thermodynamic Properties of Fuels

Typical “Hydro-Carbon” Fuels

Table 3.5 Molecular Mass, Stoichiometric Air–Fuel Ratios.

<i>Fuel</i>	Chemical Formula	<i>M</i> (kg/kmole)	<i>AF_s</i>
Hydrogen	H ₂	2.02	34.06
Methane	CH ₄	16.04	17.12
Ammonia	NH ₃	17.03	6.05
Methanol	CH ₄ O	32.04	6.43
Propane	C ₃ H ₈	44.09	15.57
Ethanol	C ₂ H ₆ O	46.07	8.94
Nitromethane	CH ₃ NO ₂	61.04	1.69
Gasoline	≈ C ₇ H ₁₇	101.21	15.27
Octane	C ₈ H ₁₈	114.22	15.03
Diesel	≈ C _{14.4} H _{24.9}	198.04	14.30
Tetradecane	C ₁₄ H ₃₀	198.39	14.54
Hexadecane (Cetane)	C ₁₆ H ₃₄	226.44	14.56

Combustion Process



Stoichiometric combustion of methane with *air* is then



Thermodynamic Properties Of Combustion

Clifford Lane



Thermodynamic Properties Of Combustion:

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