

TABLE B.5.1 (continued)  
Saturated R-134a

Temp. (°C)	Press. (kPa)	Enthalpy, kJ/kg			Entropy, kJ/k-K		
		Sat. Liquid $h_f$	Evap. $h_{fg}$	Sat. Vapor $h_g$	Sat. Liquid $s_f$	Evap. $s_{fg}$	Sat. Vapor $s_g$
-70	8.3	119.47	235.15	354.62	0.6645	1.1575	1.8220
-65	11.7	123.18	234.55	357.73	0.6825	1.1268	1.8094
-60	16.3	127.53	233.33	360.86	0.7031	1.0947	1.7978
-55	22.2	132.37	231.63	364.00	0.7256	1.0618	1.7874
-50	29.9	137.62	229.54	367.16	0.7493	1.0286	1.7780
-45	39.6	143.18	227.14	370.32	0.7740	0.9956	1.7695
-40	51.8	148.98	224.50	373.48	0.7991	0.9629	1.7620
-35	66.8	154.98	221.67	376.64	0.8245	0.9308	1.7553
-30	85.1	161.12	218.68	379.80	0.8499	0.8994	1.7493
-26.3	101.3	165.80	216.36	382.16	0.8690	0.8763	1.7453
-25	107.2	167.38	215.57	382.95	0.8754	0.8687	1.7441
-20	133.7	173.74	212.34	386.08	0.9007	0.8388	1.7395
-15	165.0	180.19	209.00	389.20	0.9258	0.8096	1.7354
-10	201.7	186.72	205.56	392.28	0.9507	0.7812	1.7319
-5	244.5	193.32	202.02	395.34	0.9755	0.7534	1.7288
0	294.0	200.00	198.36	398.36	1.0000	0.7262	1.7262
5	350.9	206.75	194.57	401.32	1.0243	0.6995	1.7239
10	415.8	213.58	190.65	404.23	1.0485	0.6733	1.7218
15	489.5	220.49	186.58	407.07	1.0725	0.6475	1.7200
20	572.8	227.49	182.35	409.84	1.0963	0.6220	1.7183
25	666.3	234.59	177.92	412.51	1.1201	0.5967	1.7168
30	771.0	241.79	173.29	415.08	1.1437	0.5716	1.7153
35	887.6	249.10	168.42	417.52	1.1673	0.5466	1.7139
40	1017.0	256.54	163.28	419.82	1.1909	0.5214	1.7123
45	1160.2	264.11	157.85	421.96	1.2145	0.4962	1.7106
50	1318.1	271.83	152.08	423.91	1.2381	0.4706	1.7088
55	1491.6	279.72	145.93	425.65	1.2619	0.4447	1.7066
60	1681.8	287.79	139.33	427.13	1.2857	0.4182	1.7040
65	1889.9	296.09	132.21	428.30	1.3099	0.3910	1.7008
70	2117.0	304.64	124.47	429.11	1.3343	0.3627	1.6970
75	2364.4	313.51	115.94	429.45	1.3592	0.3330	1.6923
80	2633.6	322.79	106.40	429.19	1.3849	0.3013	1.6862
85	2926.2	332.65	95.45	428.10	1.4117	0.2665	1.6782
90	3244.5	343.38	82.31	425.70	1.4404	0.2267	1.6671
95	3591.5	355.83	64.98	420.81	1.4733	0.1765	1.6498
100	3973.2	374.74	32.47	407.21	1.5228	0.0870	1.6098
101.2	4064.0	390.98	0	390.98	1.5658	0	1.5658

# Thermodynamics Tables In Si Units

**A Gutmann**



## **Thermodynamics Tables In Si Units:**

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**Wasserdampftafeln** U. Grigull, J. Straub, P. Schiebener, 2012-12-06 The tables and diagrams concerning the properties of ordinary water substance as offered in this booklet are mainly meant for use by students at universities and colleges so that they may be able to solve problems in the fields of power and chemical engineering where water and steam are serving as working or process medium On the other hand the tables and diagrams should support engineers in research work and industrial practice to obtain a quick and reliable general view of the proper ties of water substance The thermodynamic properties of state have been calculated according to a formu lation given by Haar Gallagher and Kell this formulation was preliminarily adopted in 1983 by the International Association for the Properties of Steam IAPS All the other properties have been calculated according to the respective Releases of IAPS Only units of the International System of Units SI Units and their decimal multi ples and parts have been used The detailed conversion tables facilitate comparisons with former material We hope that the Student s Tables will prove a useful source for both students and engineers Munich May 1984 The Editors Vorwort Die hier vorgelegten Tafeln und Diagramme uber die Eigenschaften von gew6hnlichem Wasser sind in erster Linie fur den Gebrauch der Studenten an Universitaten und Fach hochschulen bestimmt Diese sollen dam it Probleme aus der Energietechnik und der Verfahrenstechnik 16sen k6nnen bei denen Wasser und Wasserdampf als Arbeits oder ProzeBmedium eine Rolle spielen **Handbook of Thermodynamic Tables and Charts** K. Raznjevic, 1976 This volume provides comprehensive treatment of basic data for engineering thermodynamics as well as practical and theoretical thermodynamics All numerical values are cited in both technical and SI units Contents are arranged according to aggregate states solids liquids vapors and gases And at the end of the book is a table converting units of measure from one system to another Pref Information on thermal properties melting points specific heat thermal conductivity etc Twenty one pages of conversion factors and equivalents Two large charts in inside back cover on water enthalpy vs entropy Indexed Published 1976 [Thermodynamic Tables to Accompany Modern Engineering Thermodynamics](#) Robert Balmer, 2011-01-25 Thermodynamic Tables to Accompany Modern Engineering Thermodynamics is a companion text to Modern Engineering Thermodynamics by Robert T Balmer It contains two Appendices Appendix C features 40 thermodynamic tables while Appendix D provides 6 thermodynamic charts These charts and tables are provided in a separate booklet to give instructors the flexibility of allowing students to bring the tables into exams This booklet is provided at no extra charge with new copies of Balmer s book It may be purchased separately if needed **TRC Thermodynamic Tables, Hydrocarbons**, 1986

**Modern Engineering Thermodynamics - Textbook with Tables Booklet** Robert T. Balmer, 2011-01-03 Modern

Engineering Thermodynamics Textbook with Tables Booklet offers a problem solving approach to basic and applied engineering thermodynamics with historical vignettes critical thinking boxes and case studies throughout to help relate abstract concepts to actual engineering applications It also contains applications to modern engineering issues This textbook is designed for use in a standard two semester engineering thermodynamics course sequence with the goal of helping students develop engineering problem solving skills through the use of structured problem solving techniques The first half of the text contains material suitable for a basic Thermodynamics course taken by engineers from all majors The second half of the text is suitable for an Applied Thermodynamics course in mechanical engineering programs The Second Law of Thermodynamics is introduced through a basic entropy concept providing students a more intuitive understanding of this key course topic Property Values are discussed before the First Law of Thermodynamics to ensure students have a firm understanding of property data before using them Over 200 worked examples and more than 1 300 end of chapter problems provide an extensive opportunity to practice solving problems For greater instructor flexibility at exam time thermodynamic tables are provided in a separate accompanying booklet University students in mechanical chemical and general engineering taking a thermodynamics course will find this book extremely helpful Provides the reader with clear presentations of the fundamental principles of basic and applied engineering thermodynamics Helps students develop engineering problem solving skills through the use of structured problem solving techniques Introduces the Second Law of Thermodynamics through a basic entropy concept providing students a more intuitive understanding of this key course topic Covers Property Values before the First Law of Thermodynamics to ensure students have a firm understanding of property data before using them Over 200 worked examples and more than 1 300 end of chapter problems offer students extensive opportunity to practice solving problems Historical Vignettes Critical Thinking boxes and Case Studies throughout the book help relate abstract concepts to actual engineering applications For greater instructor flexibility at exam time thermodynamic tables are provided in a separate accompanying booklet

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