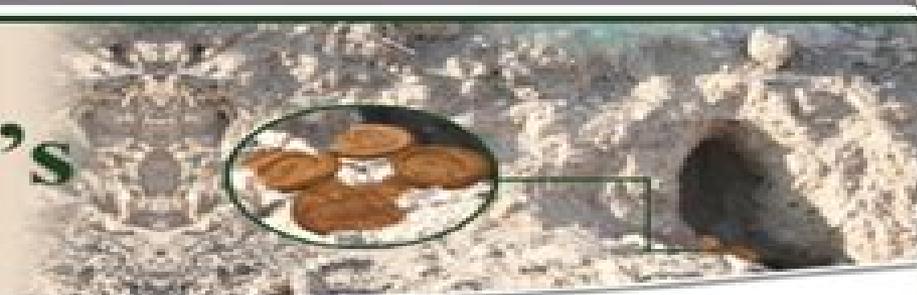
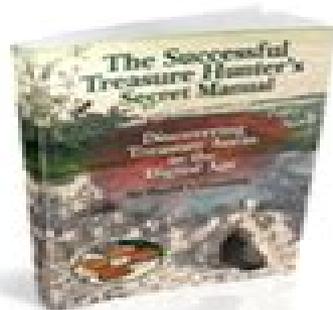


The Successful Treasure Hunter's Secret Manual



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The Successful Treasure Hunter's Secret Manual



I was totally amazed when I saw the brilliant red aura, right over the buried gold...

Dear fellow treasure hunter

THE MOST SOUGHT AFTER INFORMATION ON FINDING BURIED TREASURE CAN NOW BE YOURS

We both know the problem, you get a good lead on a buried cache, which turns out to be somewhere in hundreds of acres of land. How do you find it? Well you could spend years searching every square inch with a metal detector, providing the terrain is suitable but, if the treasure is too deep for your detector, you would NEVER find it. Then again, you could spend big bucks on controversial long-range locators, which at best only hint at the location or you could try dowsing, which is at least free.

Louis J Matacia came up with an answer in the form of a certain Polaroid camera able to photograph treasure auras, which was great, I found treasure myself using just such a tool. The problem now is that the camera is obsolete, although still fairly readily available on the used market and Polaroid stopped making the film in 2005. The last original film pack I bought was two years out of date and cost \$60. And it will only get worse from here on in, eventually becoming unobtainable at any price! So is that the end of using auras to locate buried treasure?

Not at all I set about trying to find a solution and came up with two! The first is that there is a widely used Polaroid film readily available that can easily be used in the camera. And more importantly it captures treasure auras just as good or better than the original film. The second solution discussed in great detail is – wait for it – YOU CAN CAPTURE TREASURE AURAS WITH DIGITAL CAMERAS! Not collector's items like the Polaroid, mind you, but bang up to date models that you can buy anywhere tomorrow...

AMAZING NEW DISCOVERY LETS YOU SEE BURIED TREASURE FROM A DISTANCE

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Treasure Hunters Manual 7

JS Bruner



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web the six NO_3^- ions and the six Na^+ ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation $3\text{Ba}^{2+} + 2\text{PO}_4^{3-} \rightarrow \text{Ba}_3(\text{PO}_4)_2$

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