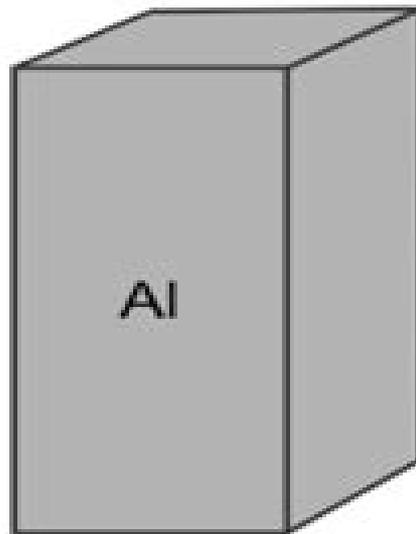


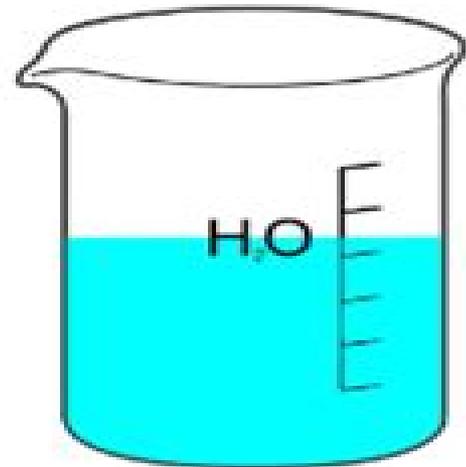
copper block

specific heat
of copper
= $390 \text{ J/kg}^\circ\text{C}$



aluminum block

specific heat
of aluminum
= $910 \text{ J/kg}^\circ\text{C}$



water

specific heat
of water
= $4200 \text{ J/kg}^\circ\text{C}$



Lower Specific Heat Capacity -

warms up and cools down quickly
as it takes much less energy to
change its temperature

Higher Specific Heat Capacity -

warms up and cools down slowly
as it takes up much more energy
to change its temperature

Specific Heat Of Solids

Lingjun Ying



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