

# Simple Health

A Straightforward Guide to Feeling Better Every Day

- Sleep
- Food
- Movement
- Stress
- Mindset
- Habits

One small step at a time

# Straightforward Guide To Alternative Health

**JE Gale**



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14 7 3 14  $\Delta h_{rxn} = q_{rxn} / c_{calorimeter} m c_s \Delta t$

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web in this set of practice questions we will go over the main types of questions on calorimetry including the heat capacity

the heat of reaction finding the final temperature of a mixture constant pressure calorimetry and constant volume calorimetry a few important concepts and formulas you will need to solve these questions

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web calorimetry worksheet 1 if 0.315 moles of hexane  $C_6H_{14}$  is combusted in a bomb calorimeter containing 5.65 liters of water calculate the molar heat of combustion of hexane if the water temperature rises 5.54 °C the specific heat capacity of water is 4.184 J/g °C

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web 6 a 218 g sample of steam at 121 °C is cooled to ice at 14 °C find the change in heat content of the system 7 in going from ice at 34 °C to steam at 138 °C a sample of water absorbs  $1.41 \times 10^5$  J find the mass of the sample answers 1 2 68 x 10<sup>4</sup> J 2 6 23 x 10<sup>4</sup> J 3 1 71 x 10<sup>4</sup> J 4 1 11 x 10<sup>5</sup> J 5 2 11 x 10<sup>5</sup> J 6

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