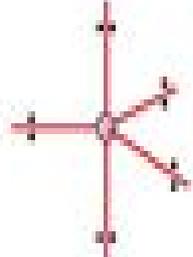
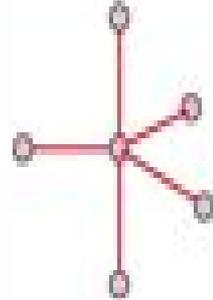
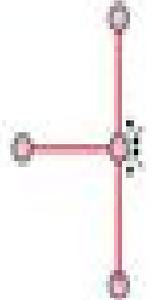
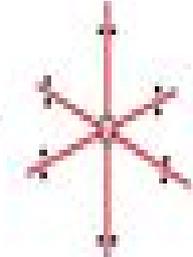
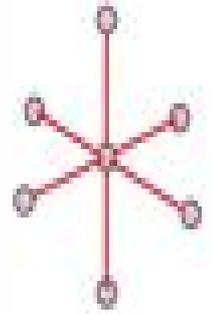
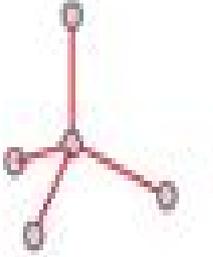
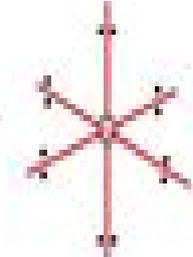
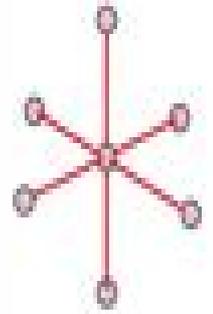
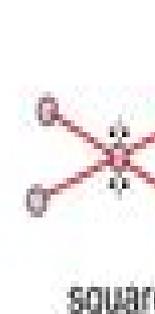


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number and arrangement of electron pairs	molecular shapes	number and arrangement of electron pairs	molecular shapes
<p>2  linear</p>	<p> linear</p>	<p>5  trigonal bipyramidal</p>	<p> trigonal bipyramidal</p> <p> seesaw</p> <p> T-shaped</p>
<p>3  trigonal planar</p>	<p> trigonal planar</p> <p> angular</p>	<p>4  tetrahedral</p>	<p> tetrahedral</p> <p> trigonal pyramidal</p> <p> angular</p>
<p>4  tetrahedral</p>	<p> tetrahedral</p> <p> trigonal pyramidal</p> <p> angular</p>	<p>6  octahedral</p>	<p> octahedral</p> <p> square pyramidal</p> <p> square planar</p>

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