

CONSTANTES, UNITÉS ET SYMBOLES USUELS

Quelques constantes fondamentales (IUPAC 2015)			
Grandeur physique	Symbole	Valeur	Unités
constante d'Avogadro	N_A	$6,02214179 \cdot 10^{23}$	mol^{-1}
vitesse de la lumière dans le vide	c, c_0	$2,99792458 \cdot 10^8$	$\text{m} \cdot \text{s}^{-1}$
accélération de la pesanteur	g	9,80665	$\text{m} \cdot \text{s}^{-2}$
charge élémentaire	e	$1,602176487 \cdot 10^{-19}$	C
volume molaire du gaz parfait à T.P.N.	V_m	$2,241410 \cdot 10^{-2}$	$\text{m}^3 \cdot \text{mol}^{-1}$
et sous 1 bar à 273,15 K	V_m^0	$2,271108 \cdot 10^{-2}$	$\text{m}^3 \cdot \text{mol}^{-1}$
constante des gaz	R	8,314472	$\text{J} \cdot \text{K}^{-1} \cdot \text{mol}^{-1}$
	R	0,08205783	$\text{L} \cdot \text{atm} \cdot \text{K}^{-1} \cdot \text{mol}^{-1}$
constante de Boltzmann	k, k_B	$1,3806504 \cdot 10^{-23}$	$\text{J} \cdot \text{K}^{-1}$
constante de Planck	h	$6,62606896 \cdot 10^{-34}$	J·s
constante de Faraday	F	$9,64853399 \cdot 10^4$	$\text{C} \cdot \text{mol}^{-1}$
constante de Rydberg	R_∞	$1,0973731534 \cdot 10^7$	m^{-1}
		$2,1798741 \cdot 10^{-18}$	J
unité de masse atomique	m_u, u	$1,6605402 \cdot 10^{-27}$	kg
masse de l'électron au repos	m_e	$9,1093897 \cdot 10^{-31}$	kg
masse du proton au repos	m_p	$1,6726231 \cdot 10^{-27}$	kg
masse du neutron au repos	m_n	$1,6749286 \cdot 10^{-27}$	kg
rayon de Bohr	a_0	$5,29177249 \cdot 10^{-11}$	m
perméabilité du vide	μ_0	$4\pi \cdot 10^{-7}$	$\text{N} \cdot \text{A}^{-2}$
permittivité du vide	ϵ_0	$8,854187817 \cdot 10^{-12}$	$\text{F} \cdot \text{m}^{-1}$

Alphabet grec					
Nom	minuscule	majuscule	Nom	minuscule	majuscule
Alpha	α	A	nu	ν	N
bêta	β	B	xi	ξ	X
gamma	γ	Γ	omicron	\omicron	O
delta	δ	Δ	pi	π	P
epsilon	ϵ	E	rho	ρ	R
zêta	ζ	Z	sigma	σ	S
êta	η	H	tau	τ	T
thêta	θ	Θ	upsilon	υ	Y
iota	ι	I	phi	ϕ	Φ
kappa	κ	K	chi	χ	X
lambda	λ	Λ	psi	ψ	Ψ
mu	μ	M	oméga	ω	Ω

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National Research Council (U.S.)



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Tables annuelles de constantes et données numériques de chimie, de physique et de technologie, 1921 *International Critical Tables of Numerical Data, Physics, Chemistry and Technology* National Research Council (U.S.), 1930

International Critical Tables of Numerical Data, Physics, Chemistry and Technology, 1928 Tables for the Calculation of Tides by Means of Harmonic Constants International Hydrographic Organization, 1926 **Metals** Simonne

Allard, 2013-10-22 *Metals Thermal and Mechanical Data* covers the thermal and mechanical data of metals The book presents topics on atomic heat phase changes and vapor pressure and elastic constants The text also includes topics on plasticity properties such as stress strength activation energy for and change in creep and hardness internal friction and data for liquid metals close to the melting point including viscosity and surface tension Engineers technicians physical chemists as well as specialists in metallurgy will find the book invaluable NIST Serial Holdings National Institute of Standards and

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