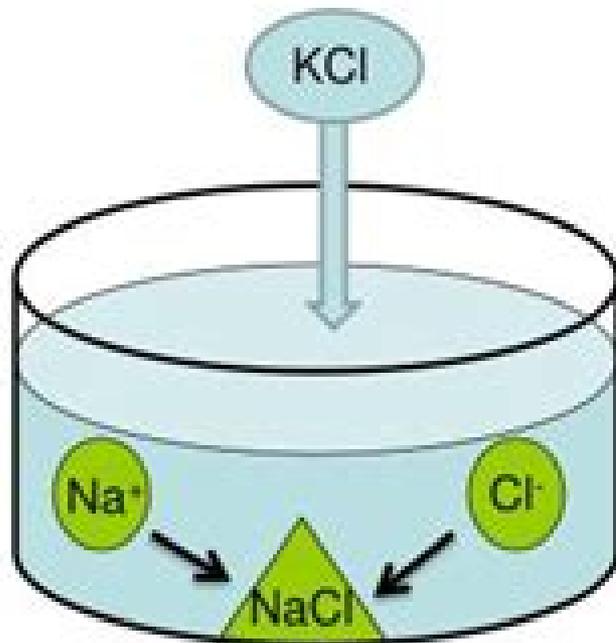


Common Ion Effect

- With solutions that are saturated, addition of either ion, from another compound, will shift the equilibrium



To a saturated NaCl solution, add KCl
Cl⁻ concentration goes up
Equilibrium shifts left
Some NaCl will precipitate
Na⁺ concentration goes down

The Ion Effect

**Dinesh Khattar, Ravi Raj Dudeja, K.K.
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The Ion Effect Fred Soyka, Alan Edmonds, 1978 *The Ion Effect* Fred Soyka, 1981 **The Ion Effect** Fred Soyka, Alan Denis Edmonds, 1977 *CBSE Chemistry Chapterwise Case Study Class 11* Priti Singhal, 2024-11-17 This book is structured to align with the latest syllabus and curriculum guidelines ensuring that the content is both relevant and rigorous Each chapter begins with a clear set of learning objectives providing a roadmap for students to understand what they will achieve by the end of the chapter We have included numerous diagrams illustrations and real life examples to make complex concepts more accessible and engaging

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