

SOME THERMODYNAMIC RELATIONS

HEAT TRANSFER

Conduction $Q = kA dt/dx$ k is the thermal conductivity

Convection

Radiation $P = \epsilon\sigma T^4$ per unit area. ϵ is the emissivity,
 σ is the Stefan-Boltzmann constant
 $= 5.67 \cdot 10^{-8} \text{ W m}^{-2} \text{ K}^{-4}$

1st Law $\Delta U = \bar{d}Q - \bar{d}W$ For fluids $\bar{d}W = pdV$

Entropy $\Delta S = \Delta Q_{REV} / T$

Efficiency of a Carnot Engine $\eta_C = 1 - T_2 / T_1$

Enthalpy $h = u + pv$ Useful for describing

1. Heat transfer at constant pressure in a boiler or
condenser. Heat input $Q = h_2 - h_1$ (2.1)

2. Adiabatic changes in compressors and turbines where
net work on the shaft $W = h_1 - h_2$ (2.2)

(See derivation 2.2 in the text book for proof of these last two relations)

Thermodynamic Relations In Multi Compone

Sabine Zange



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