

When a solute dissolves three processes occur simultaneously:

- The bonds holding the molecules or ions together are broken. This process requires energy (**endothermic**).
- Some of the intermolecular forces between the particles of the solvent also break (**endothermic**).
- The molecules or ions of the solute become attracted by the molecules of the solvent and new bonds are formed. This process releases energy (**exothermic**). The solute particles are dispersed evenly throughout the solution.

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