

Valency of First 20 Elements

Element Name	Symbol	Atomic Number	Electronic Configuration	Lose/gain/share Electrons	Valency
Hydrogen	H	1	1	Lose or share 1 Electron	1
Helium	He	2	2	Stable as per Duplet rule	0
Lithium	Li	3	2,1	Lose 1 Electrons	1
Beryllium	Be	4	2,2	Lose 2 Electrons	2
Boron	B	5	2,3	Lose 3 Electrons	3
Carbon	C	6	2,4	Share 4 Electrons	4
Nitrogen	N	7	2,5	Gain or share 3 Electrons	3
Oxygen	O	8	2,6	Gain or share 2 Electrons	2
Fluorine	F	9	2,7	Gain or share 1 Electrons	1
Neon	Ne	10	2,8	Stable as per Octet rule	0
Sodium	Na	11	2,8,1	Lose 1 Electrons	1
Magnesium	Mg	12	2,8,2	Lose 2 Electrons	2
Aluminum	Al	13	2,8,3	Lose 3 Electrons	3
Silicon	Si	14	2,8,4	Share 4 Electrons	4
Phosphorous	P	15	2,8,5	Gain or share 3 Electrons	3
Sulfur	S	16	2,8,6	Gain or share 2 Electrons	2
Chlorine	Cl	17	2,8,7	Gain or share 1 Electrons	1
Argon	Ar	18	2,8,8	Stable as per Octet rule	0
Potassium	K	19	2,8,8,1	Lose 1 Electrons	1
Calcium	Ca	20	2,8,8,2	Lose 2 Electrons	2

Valency Primer

F. R. Hartley



Valency Primer:

A Valency Primer James Clare Speakman, 1968 **A valency primer** James Clare Speakman, *A Valency Primer* James Clare Speakman, 1968 A Primer on Riemann Surfaces A. F. Beardon, 1984-10-18 An Introduction to Crystal Chemistry R. C. Evans, 1966 First published in 1964 as the second edition of a 1939 original this well known textbook presents the fundamental principles of crystal chemistry at a level that was suitable for undergraduate students of chemistry physics metallurgy mineralogy and related subjects at the time of its publication The first part deals with the general principles of crystal architecture in terms of predominant types of binding forces between the atoms themselves There are chapters on atomic structure and the ionic covalent metallic and van der Waals bonds The second part contains a discussion of systematic crystal chemistry in which the physical and chemical properties of crystalline substances are related to their structures Selective Toxicity Adrien Albert, 2012-12-06 This book is about selectively toxic agents That is to say it is about those substances that affect certain cells without harming others even when they are close neighbours Toxicity need not be fatal It can be made easily reversible as is the case with general anaesthetics Selective toxicity covers an immense field most of the drugs used for treating illness in man and his economic animals as well as all of the fungicides insecticides and weed killers that are used in agriculture Essentially this book is a discussion of the physical and chemical means which contribute to selectivity and this is the basis of molecular pharmacology _Selective Toxicity began as a course of lectures that Professor F G Young encouraged me to give in University College London in 1948 and again in 1949 The first edition appeared in 1951 as a very small book because little was then known about the factors that provide selectivity Since those early days the subject has undergone tremendous development At first industry was un receptive to the word toxicity however qualified Yet the market was being supplied with biologically powerful substances of which several had the potential to cause harm This aspect was brought to light by two events of the early 1960s The first of these was the discovery that a sedative thalidomide administered to expectant mothers after what was then considered to be adequate testing had caused permanent deformities in about 10000 children Eosinophils in Health and Disease James J. Lee, Helene F. Rosenberg, 2012-11-02 Eosinophils in Health and Disease provides immunology researchers and students with a comprehensive overview of current thought and cutting edge eosinophil research providing chapters on basic science disease specific issues therapeutics models for study and areas of emerging importance *Journal of Protective Coatings & Linings*, 1995 **Handbook of Surface and Colloid Chemistry** K. S. Birdi, 1997-10-22 European North American Canadian and South Asian scientists have joined forces to create the only handbook in existence on the chemistry of surface and colloidal systems Never before has the massive amount of data required by surface research chemists been available in a single volume With this new handbook searching through journals for a piece of data becomes obsolete All the facts and figures you need in the laboratory or in the classroom are at your finger tips The data is presented in a unique style and format providing a guide for future research planning

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